

DPM 957-1

Material Safety

Electro Chemicals Division
Diamond Shamrock Corporation
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Diamond Shamrock

GENERAL INFORMATION

Liquid Caustic Soda is a clear, corrosive liquid with no distinct odor.

Data Sheet

340B

I PRODUCT IDENTIFICATION

MANUFACTURER'S NAME DIAMOND SHAMROCK CORPORATION	REGULAR TELEPHONE NO. EMERGENCY TELEPHONE NO.	Contact Local Sales Office 216-357-7070
ADDRESS Divisional Technical Center, P. O. Box 191, Painesville, Ohio 44077		
TRADE NAME CAUSTIC SODA - Liquid		
SYNONYMS Sodium Hydroxide; NaOH		

II HAZARDOUS INGREDIENTS

MATERIAL OR COMPONENT	%	HAZARD DATA
SODIUM HYDROXIDE	50	PEL* = 2.0 mg/m ³ for 15 minutes
* OSHA Permissible Exposure Limit (PEL)		

III PHYSICAL DATA

BOILING POINT, 760 MM HG 143°C	MELTING POINT	FREEZING POINT 12.1°C (54°F)
SPECIFIC GRAVITY (H ₂ O=1) 1.54 @ 15.6°C	VAPOR PRESSURE 13 mm Hg @ 60°C	
VAPOR DENSITY (AIR=1) Not applicable	SOLUBILITY IN H ₂ O, % BY WT. Completely soluble	
% VOLATILES BY VOL. Not volatile	EVAPORATION RATE (BUTYL ACETATE=1) Does not apply	
APPEARANCE AND ODOR Clear - no odor		
pH 7.5% solution has pH 14		

EC-LC-6c

All information, recommendations and suggestions appearing herein concerning our product are based upon tests and data believed to be reliable, however, it is the user's responsibility to determine the safety, toxicity, and suitability for his own use of the product described herein. Since the actual use by others is beyond our control, no guarantee, expressed or implied, is made by Diamond Shamrock Corporation as to the effects of such use, the results to be obtained, or the safety and toxicity of the product nor does Diamond Shamrock Corporation assume any liability arising out of use, by others, of the product referred to herein. Nor is the information herein to be construed as absolutely complete. Additional test information may be necessary or desirable when particular or exceptional conditions or circumstances exist or because of applicable laws or government regulations.

BOE-C6-0209918

IV FIRE AND EXPLOSION DATA		
FLASH POINT (TEST METHOD) None		AUTOIGNITION TEMPERATURE Nonflammable
FLAMMABLE LIMITS IN AIR, % BY VOL	LOWER Nonflammable	UPPER Nonflammable
EXTINGUISHING MEDIA Use carbon dioxide, "alcohol" foam or dry chemicals in areas where caustic soda is stored. Caustic soda is nonflammable.		
SPECIAL FIRE FIGHTING PROCEDURES Pressure-demand, self-contained respiratory protection and protective clothing should be worn by firefighters in areas where caustic soda is stored. Caustic soda is nonflammable.		
UNUSUAL FIRE AND EXPLOSION HAZARD None		
V HEALTH HAZARD INFORMATION		
HEALTH HAZARD DATA PEL = 2.0 mg/m ³ for 15 minutes. Acute LC ₅₀ > 0.018 < 0.20 mg/L; Acute LD ₅₀ = 140-340 mg/kg (oral - rat) Acute LD ₅₀ = 1,350 mg/kg (dermal - rabbit)		
ROUTES OF EXPOSURE Caustic soda is a corrosive material.		
INHALATION Airborne concentrations of dust, mist, or spray of caustic soda may cause damage to the upper respiratory tract and even to the lung tissue proper which could produce chemical pneumonia, depending upon severity of exposure.		
SKIN CONTACT Caustic soda is destructive to tissues contacted and produces severe burns.		
SKIN ABSORPTION See "Skin Contact" above.		
EYE CONTACT Caustic soda is destructive to eye tissues on contact. Will cause severe burns that result in damage to the eyes and even blindness.		
INGESTION Caustic soda can cause severe burns and complete tissue perforation of mucous membranes of the mouth, throat, esophagus, and stomach if swallowed.		
EFFECTS OF OVEREXPOSURE ACUTE OVEREXPOSURE Burns: resulting in frequently deep ulceration and ultimate scarring.		
CHRONIC OVEREXPOSURE The chronic local effect may consist of multiple areas of superficial destruction of the skin or of primary irritant dermatitis. Similarly, inhalation of spray or mist may result in varying degrees of irritation or damage to the respiratory tract tissues.		
EMERGENCY AND FIRST AID PROCEDURES Object is to Seek Medical Attention Immediately.		
EYES Immediately flush eyes with large amounts of water for at least 15 minutes holding eyelids apart to ensure flushing of the entire eye surface. Washing eyes within 1 minute is essential to achieve maximum effectiveness. Seek medical attention immediately.		
SKIN Immediately wash contaminated skin with plenty of water. This may be followed with a rinse with vinegar or dilute acetic acid (3% solution) if available. Remove contaminated clothing and footwear and wash clothing before reuse. Discard footwear which cannot be decontaminated. Seek medical attention immediately.		
INHALATION Get person out of contaminated area to fresh air. If breathing has stopped, artificial respiration should be started. Oxygen may be administered, if readily available. Seek medical attention immediately.		
INGESTION If swallowed, DO NOT induce vomiting. Give large quantities of water. If available, give several glasses of milk. Never give anything by mouth to an unconscious person. Seek medical attention immediately.		
NOTES TO PHYSICIAN		

Data Sheet

VI REACTIVITY DATA**CONDITIONS CONTRIBUTING TO INSTABILITY**

Under normal use conditions, caustic soda is stable.

INCOMPATIBILITY

When handling caustic soda, avoid contact with aluminum, leather, wool, tin, zinc, and alloys containing these metals. Do not mix with strong acids without dilution and agitation to prevent violent or explosive reaction.

HAZARDOUS DECOMPOSITION PRODUCTS

None

CONDITIONS CONTRIBUTING TO HAZARDOUS POLYMERIZATION

None

VII SPILL OR LEAK PROCEDURES**STEPS TO BE TAKEN IF MATERIAL IS RELEASED OR SPILLED**

Stop leaks. Contain spill. Remove as much as possible (e.g., shovel up or remove by vacuum truck, if liquid). Neutralize remaining traces of material with dilute acid; then flush area with water followed by liberal covering of sodium bicarbonate. Reuse spilled material, if possible, otherwise place in a closed, labeled container, and store in a safe place to await proper disposal. Persons performing this work should wear adequate personal protective equipment and clothing. **Caution:** Caustic soda may react violently with acids and water.

NEUTRALIZING CHEMICALS

Neutralize with any dilute inorganic acid such as hydrochloric, sulfuric, nitric, phosphoric, and acetic acid.

WASTE DISPOSAL METHOD

Dispose in accordance with all federal, state and local regulations concerning health and pollution. Dispose via approved chemical waste disposal method, if regulations permit.

VIII INDUSTRIAL HYGIENE CONTROL MEASURES**VENTILATION REQUIREMENTS**

Good industrial hygiene practice dictates that the work area should be isolated and contained, and provided with adequate local exhaust ventilation or other controls to maintain the air concentration of caustic soda below 2.0 mg/m³ as required by OSHA. Air concentration of carbon monoxide formed by reaction of caustic soda and reducing sugars should not exceed 50 ppm for an eight (8) hour TWA.

SPECIFIC PERSONAL PROTECTIVE EQUIPMENT**RESPIRATORY (SPECIFY IN DETAIL)**

Use NIOSH-approved respirator for dusts and mists. Use air purifying respirator where caustic soda is in contact with reducing sugars.

EYE

Chemical splash goggles and face shield should be worn when working with or around caustic soda.

GLOVES

Gloves coated with rubber, synthetic elastomers, PVC, or other plastic should be worn when handling caustic soda to minimize skin contact.

OTHER CLOTHING AND EQUIPMENT

Hard hats, safety shoes, and rubber boots should be worn along with rubber apron when handling caustic soda. Safety showers and eyewash stations should be provided in all areas in which caustic soda is handled.

IX SPECIAL PRECAUTIONS

PRECAUTIONARY STATEMENTS

DANGER!

Causes Severe Burns to Skin and Eyes

Do NOT get in eyes, on skin, on clothing.

Avoid breathing dust, mist, or spray.

Do NOT take internally.

Use with adequate ventilation and employ respiratory protection when exposed to dust, mist or spray.

When handling, wear chemical splash goggles, face shield, rubber gloves and protective clothing.

Wash thoroughly after handling.

Avoid contact with strong acids to prevent violent or explosive reactions.

Keep container closed.

Hazardous carbon monoxide gas can form upon contact with food and beverage products in enclosed spaces and can cause death. Follow appropriate tank entry procedures (ANSI Z117.1-1977).

First Aid:

In case of contact:

For eyes: Immediately flush with plenty of water for at least 15 minutes holding eyelids apart to ensure flushing of the entire eye surface. **Seek medical attention immediately.**

Skin: Immediately wash with plenty of water. If available, rinse with vinegar or dilute acetic acid (3% solution). Remove contaminated clothing and footwear. Wash clothing before reuse and discard footwear which cannot be decontaminated. **Seek medical attention immediately.**

Inhalation: Remove person from contaminated area to fresh air. If breathing has stopped, artificial respiration should be started. Oxygen may be administered if readily available. **Seek medical attention immediately.**

Ingestion: If swallowed, DO NOT induce vomiting. Give large quantities of water. If available, give several glasses of milk. **NEVER** give anything by mouth to an unconscious person. **Seek medical attention immediately.**

Special instructions for dissolving caustic soda:

When making solution, **always** add slowly to liquid surface with constant stirring. **Never add the liquid to the caustic soda.**

Always start with lukewarm liquid (80°-100°F.) **Never** start with hot or cold liquid.

If caustic soda becomes concentrated in one area, or if added too rapidly, or if added to hot or cold liquid, a rapid temperature increase can result in **DANGEROUS** boiling and/or spattering which may cause an immediate **VIOLENT ERUPTION**.

Spill or Leak: Leaks should be stopped. Spills, after containment, should be shoveled up and removed to chemical waste area or removed by vacuum truck, if liquid. Neutralize residue with dilute acid, flush spill area with water followed by liberal covering of sodium bicarbonate. Dispose of wash water according to Federal, State and Local regulations.

For Industrial Use Only

OTHER HANDLING AND STORAGE REQUIREMENTS

Caustic soda is classified by DOT as a corrosive material.

Considerable heat is generated when water is added to caustic soda; therefore, when making solutions **always** add the caustic soda to the water with constant stirring. The water should always be lukewarm (80°-100°F) **Never** start with hot or cold water. If caustic soda becomes concentrated in one area, or if added too rapidly, or if added to hot or cold water, a rapid temperature increase can result in **DANGEROUS BOILING** and/or spattering or may cause an immediate **VIOLENT ERUPTION**. Caustic soda can react violently or explosively with acids and many organic chemicals.

Caustic soda reacts with reducing sugars such as fructose, lactose, maltose, galactose, levulose and arabinose to form carbon monoxide. While the potential for worker exposure to carbon monoxide may be small, a potential does exist during cleaning of certain dairy and possibly other industry equipment.

Returnable containers should be shipped in accordance with supplier's recommendations. Return shipments should comply with all federal, state and DOT regulations. All residual caustic soda should be removed from containers prior to disposal.

More information on the hazards and handling of caustic soda appear in Diamond Shamrock Corporation's Caustic Soda Handbook EC-LDC-1c.

DEPARTMENT OF TRANSPORTATION INFORMATION

PROPER SHIPPING NAME: Caustic Soda, Liquid

HAZARD CLASS: Corrosive Material

I.D. NO.: UN 1824

HAZARD SUBSTANCE: RQ-1000

PREPARED BY Diamond Shamrock Corporation
Technical Service Group

DATE

May 1, 1981